PERISELECTIVE CYCLOADDITION REACTION OF 3-PHENYLTHIA^{IV} ZOLO[4,3-a]ISOINDOLE TO ELECTRON-DEFICIENT OLEFIN

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3-Phenylthia^{IV}zolo[4,3-a]isoindole, generated in situ from 3-phenyl-5H-thiazolo-[4,3-a]isoindolium bromide and triethylamine, reacted with a variety of electron-deficient olefins yielding the regio- and stereoselective 1:1 adducts between the azomethine ylide 1,3-dipole of the isoindole and the olefins.

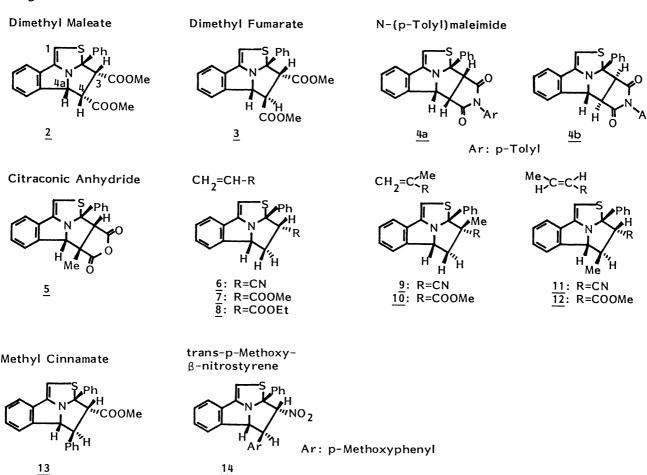
It has been recently reported that the peripheral cycloaddition reaction of azomethine ylide containing a bridgehead nitrogen atom as a central atom of the 1,3-dipole offers a convenient method for the preparation of cyclazine derivatives. Some mesoionic azapentalenes, indolizines, and their analogs exhibit similar reactions with electron-deficient acetylenes and olefins to give a variety of cyclazine derivatives, while the reaction mechanisms have not been clearly established. On the other hand, a certain nitrogen-bridged tetravalent sulfur compounds such as a thia volo 3,4-b indazole and a pyrrolo 1,2-c thia wave demonstrated the bi-perifunctional properties reacting with dimethyl acetylenedicarboxylate as azomethine ylides and with N-phenylmaleimide as thiocarbonyl ylides. However, the periselective cycloaddition reaction to the azomethine ylide 1,3-dipole of nitrogen-bridged tetravalent sulfur compound is unknown so far.

We have already reported the synthesis of 3-phenyl-5H-thiazolo[4,3-a]isoindolium bromide that might be a precursor of 3-phenylthia IV zolo[4,3-a]isoindole $\underline{1}$ by deprotonation. 10 The present communication describes the cycloaddition reactions of $\underline{1}$ with a variety of electron-deficient olefins, in which the azomethine ylide 1,3-dipole of $\underline{1}$ was periselectively engaged to yield the regio- and stereoselective cycloadducts

An equivalent suspension of 3-phenyl-5H-thiazolo[4,3-a]isoindolium bromide and dimethyl maleate in dry chloroform was warmed at 40 °C and a slight excess of triethylamine in chloroform was slowly added with stirring under nitrogen. Stirring was continued until all the insoluble materials dissolved into a homogeneous solution. The chloroform was evaporated at room temperature and the residue was

washed with ice-water to remove triethylammonium bromide, and then extracted with benzene. Evaporation of the benzene under vacuum afforded a colorless product $\underline{2}^{11}$ as a sole product in an excellent yield (Table 1).

The similar reaction with dimethyl fumarate yielded $\underline{3}$, an isomer of $\underline{2}$, also as a sole product. Both of the products $\underline{2}$ and $\underline{3}$ are found to be the 1:1 cycloadducts to an azomethine ylide \underline{B}^{12} of 3-phenylthia^{IV}zolo[4,3-a]isoindole $\underline{1}$, that was generated from the bromide by deprotonation, on the basis of the $^1\text{H-NMR}$ spectral data shown in Table 2. The configurations of olefins used should be retained in the structures of cycloadducts, $\underline{2}$ and $\underline{3}$, formed stereoselectively. The stereochemistry of $\underline{2}$ and $\underline{3}$ was mainly based on the chemical shifts of ester methyls: the ester methyl (δ 3.28 ppm) at the 4 position of $\underline{2}$ appears in a considerably high field due to the shielding effect of the fused benzene ring. 13



N-(p-Tolyl)maleimide is the only exception among the olefins employed in the cycloaddition reactions, which afforded a mixture of two isomeric 1:1 adducts <u>4a</u> and <u>4b</u> in the ratio of 6 to 4. Although it was very difficult to separate either of the adducts in a pure form through a column chromatography, their structures were determined by reading the ¹H-NMR spectrum of the mixture. Thus the product with a rather small coupling constant between 4-H and 4a-H is an endo [3 + 2] cycloadduct <u>4a</u> and the one with a larger coupling constant is an exo cycloadduct 4b (Table 2).

In contrast with the maleimide, an another cyclic dipolar phile, citraconic anhydride, gave the only isomeric cycloadduct $\underline{5}$ in the reaction with $\underline{1}$. The product $\underline{5}$ was assigned to be an endo cycloadduct with a 4-exo-methyl group on the basis of the splitting pattern of methine hydrogens (Table 2).

The formation of $\underline{5}$ as a sole product might be explained from a viewpoint of steric hindrance: the approach between $\underline{1}$ and the anhydride leading to $\underline{5}$ seems to be sterically least hindered among the all possible approaches.

With an unsymmetrical dipolarophile with one electron-withdrawing substituent, the question of regionselectivity in the cycloaddition reaction arises. Such electron-deficient olefins as acrylonitrile, methy and ethyl acrylates, methacrylonitrile, methyl methacrylate, crotononitrile, methyl crotonate, methyl cinnamate, and trans-p-methoxy- β -nitrostyrene were found to cycloadd to $\underline{1}$ in a highly peri-,

Table 1.	The	[3+	2]	Cycloadducts of	1	to	Electron-deficient Olefins.

Products	Yields (%)	mp (℃)	IR (cm ⁻¹)	MS (m/e) ^{a)}				
2	84	122-124	1730 (CO)	393*, 249				
3	67	160-161(d)	1725 (CO)	393*, 249				
4	87 (<u>4a</u> : <u>4b</u> =	6:4)	1775, 1710 (CO)					
4 5 6 7 8 9	87	133.5-135(d)	1830, 1775 (CO)	361*, 288, 249				
<u>6</u>	59	71-73	2230 (CN)	302*, 249				
7	60	140-142	1740 (CO)	335*, 249				
8	59	133-135	1730 (CO)	349*, 249				
9	73	161-164	2215 (CN)	316*, 249				
<u>10</u>	47	129-132	1727 (CO)	349*, 249				
11	50	101-103	2230 (CN)	316*, 249				
12	43	143-146	1730 (CO)	349*, 249				
11 12 13 14	46	142.5-144.5	1740 (CO)	411*, 249				
14	60	144.5-147	155, 1350 (NO ₂)	382 (M ⁺ -NO ₂), 263, 249				

a) The ion peaks with asterisk are parent peaks and the fragment ion at 249 is assigned to $\underline{1}$.

Table 2. The 1 H-NMR spectra of $2-\frac{14}{}$.

Products	1-H ^S	3-exo	3-endo	4-exo	4-endo	4a – H	J ₃₋₄	J _{4-4a}	J gem
2	5.60	H 4.39 ^d	OMe 3.61 ^S	H 3.15 ^{dd}			10.0	8.1	
2 3	5.33	H 4.12 ^d	OMe 3.59 ^S	OMe 3.81 ^s	H 3.39 ^{dd}	4.66 ^d	8.8	8.4	
<u>4a</u>	5.46	H 3.84 ^d		H 3.28 ^{dd}		5.06 ^d	8.2	9.3	
<u>4b</u>	5.62		H 4.22 ^d		H 3.08 ^{dd}	5.38 ^d	9.9	3.2	
<u>5</u>	5.56	H 3.67 ^S		Me 1.35 ⁸		4.68 ^S			1
6	5.59	H 3.66 ^{dd}		H 2.26 ^{dt}	H 1.94 ^{dt}	4.77 ^{dd}	9.4 5.2	8.8 5.8	11.9
5 6 7	5.42	н з. 80 ^t	OMe 3.84 ^s	H 2.16 ^{dd}	H 2.16 ^{dd}	4.72 ^t	7.8	7.3	1
<u>8</u>	5.43	H 3.75 ^t	OEt 4.30 ^t	H 2.17 ^t		4.73 ^t	8.1	8.0	
<u>9</u>	5.72	Me 1.23 ^S		H 2.05 ^{dd}	No.	5.21 ^{dd}		8.6 4.0	14.6
<u>10</u>	5.39	Me 0.99 ^S	OMe 3.78 ^S	H 2.61 ^{dd}		4.76 ^{dd}		9.8 5.8	12.6
11	5.40	H 3.31 ^d		Me 1.07 ^d	H 2.18 ^m	4.01 ^d	9.2	8.2	
12	5.51	H 3.59 ^d	OMe 3.99 ^S	Me 1.12 ^d	H 2.58 ^m	4.20 ^d	10.0	9.2	
13	5.41	H 3.97 ^d	OMe 3.78 ^S		H 3.58 ^{dd}	4.56 ^d	9.2	8.8	1
11 12 13 14	5.56	н 5.80 ^d			н з.88 ^t	4.72 ^d	8.5	8.5	

a) All the spectra were taken in CDCl $_{\rm 3}$ (δ ppm) and the coupling constants are expressed in Hz.

regio- and yet stereoselective manner to give the corresponding cycloadducts $\underline{6-14}$. Regardless of the structures of olefins, all the products are endo [3 + 2] cycloadducts to an azomethine ylide \underline{B} with electron-withdrawing substituents at their 3-endo positions. The structural elucidation of them were accomplished on the basis of the spectral data shown in Table 1 and 2, especially of the 1 H-NMR spectra.

Regio- and stereochemistry of the cycloadducts $\underline{6-14}$ coincides with the orientations derived from the combination of polarized structures of olefins with \underline{B} , and also from an attractive interaction between the electron-withdrawing groups and the thiazole ring of $\underline{1}$ as has been demonstrated in the reaction with dimethyl fumarate. These approaches leading to the regionselective endo cycloadducts are not sterically hindered either.

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- 11. This experiment offers the typical procedure for the reactions of 1 with electron-deficient olefins.
- 12. According to the inspection using a Dreiding model, in the [3 + 2] cycloadduct to a thiocarbonyl ylide \underline{C} of $\underline{1}$, if an endo or an exo isomer, the bridgehead hydrogen should have exhibited a very small vicinal coupling.
- 13. The inspection of a Dreiding model shows that it seems rather difficult to estimate the dihedral angles, hence vicinal coupling constants, between the 3-H and 4-H and between the 4-H and 4a-H since the skeleton of the cycloadduct is not tightly fixed. The structures of 2 and 3 are also supported by the consideration of reaction modes in which the other olefins except for the maleimide were participated.

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